



# C. U. SHAH UNIVERSITY, WADHWAN CITY.

Faculty of: **Sciences and Life Sciences**

Course: **Bachelor of Science (Microbiology)**

Semester: **II**

Subject Code: **MDC202-1C**

Subject Name: **Fundamentals of Spectroscopy**

Sr. No	Category	Subject Code	Subject Name	Teaching hours/ Week			Credit hours	Credit Points	Evaluation Scheme/ Semester								Total
				Th	Tu	Pr			Theory				Tutorial / Practical				
									Continuous and Comprehensive Evaluation		End Semester Exams		Internal Assessment		End Semester Exams		
									Marks	Marks	Marks	Duration	Marks	Duration	Marks	Duration	
4	MDC	MDC202-1C	Fundamentals of Spectroscopy	3	-	2	5	4	10	Assignment	50	2	25	1	-	-	100

## AIM :

- The aim is to enable students to acquire a specialized understanding of how light interacts with molecules and materials.
- Different methods of optical spectroscopy and their use to examine chemical and physical properties are addressed at an advanced level.

## COURSE CONTENTS

### Course Outline for Theory

UNIT	COURSE CONTENT	TEACHING HOURS
I	<b>Definition of the spectrum</b> - Electromagnetic radiation - quantization of different forms of energies in molecules (translational, rotational, vibrational, and electronic) - Born Oppenheimer approximation. <b>Microwave Spectroscopy</b> - theory of microwave spectroscopy - selection rule - Calculation of moment of inertia and bond length of diatomic molecules.	9
II	<b>UV - Visible Spectroscopy</b> - Absorption laws. Calculations involving Beer Lambert's law - instrumentation - photo colorimeter and spectrophotometer- block diagrams with description of components - theory - types of electronic transitions - chromophore and auxochromes - Absorption bands and intensity -factors governing absorption maximum and intensity	9
III	<b>I. R. Spectroscopy</b> – principle - modes of vibration of diatomic, triatomic linear (CO <sub>2</sub> ), and nonlinear triatomic molecules (H <sub>2</sub> O) - stretching and bending vibrations - selection rules. Expression for vibrational frequency (derivation not needed).	9
IV	<b>X-ray Diffraction:</b> Diffraction geometry: Bragg's law, Diffraction Intensity: Scattering from atoms, from the contents of a unit cell; structure factor function, Application to polycrystal diffraction: powder diffraction and crystal structure determination., Diffractometer measurements	9

V	<b>Basics of TEM</b> TEM instrumentation: electron sources; electromagnetic lenses; geometric and wave optics applied to TEM; lens aberrations and resolution, Interaction between fast electron and thin crystal (TEM sample), TEM sample preparation	9

### Course Outline for Practical

SR. NO	COURSE CONTENT	Hrs.
1	Applications and Uses of Electromagnetic Radiation in the real world.	30
2	Demonstrative experiments on Microwave Spectroscopy	
3	Demonstrative experiments on UV Visible Spectroscopy	
4	Demonstrative experiments of IR Spectroscopy	
5	Demonstrative experiments on X-ray diffractions	
6	Demonstrative experiments of TEM	

### TEACHING METHODOLOGY:

- Conventional method (classroom blackboard teaching)
- ICT Techniques
- Teaching through the classroom, laboratory work
- variety of learning styles and tools (PowerPoint presentations, audio-visual resources, e-resources, seminars, workshops, models)

### LEARNING OUTCOME:

- Basic understanding of light as electromagnetic radiation, their parameters, and interaction with matter
- To learn about various spectroscopy and their applications in the real world.
- Understanding various parts of instruments, sampling methods, and analysis in given spectroscopic techniques

### Arrangement of lectures duration and practical session as per defined credit numbers:

Units	Lecture Duration (In Hrs.)		Calculation of Credits (In Numbers)		Total Lecture Duration	Credit Calculation
	Theory	Practical	Theory	Practical	Theory+ Practical	Theory+ Practical
Unit – 1	06	06	3	1	45+30	3 +1
Unit – 2	06	06				
Unit – 3	06	06				
Unit – 4	06	06				
Unit – 5	06	06				
<b>TOTAL</b>	<b>45</b>	<b>30</b>	<b>3</b>	<b>1</b>	<b>75</b>	<b>4</b>

### Evaluation:

Theory Marks	Practical Marks	Total Marks
50	00	50

## REFERENCE BOOKS:

- |    |  |   |
|----|--|---|
| 1  | Elements of Analytical Chemistry                     | R. Gopalan, P.S. Subramanian, K. Rengarajan                     |
| 2  | Fundamentals of Analytical Chemistry                 | D.A. Skoog and D.M. West  |
| 3  | Principles of Instrumental Methods of Analysis       | D.A Skoog and Saunders  |
| 4  | Instrumental Methods of Analysis                     | Willard Merit Dean and Settle                                   |
| 5  | "Elements of X-Ray Diffraction                       | Cullity, B.D., and Stock, R. S                                  |
| 6  | Advanced Techniques for Materials Characterization   | Tyagi, A.K., Roy, Mainak, Kulshreshtha, S.K., and Banerjee, S., |
| 7  | Molecular Spectroscopy                               | Jeanne J. McHale  |
| 8  | X-Ray diffraction-A practical approach               | C. Suryanarayana and M. Grant Norton                            |
| 9  | Spectroscopy: Fundamentals and Data Interpretation   | N K Fuloria, S Fuloria  |
| 10 | In Situ Transmission Electron Microscopy Experiments | Renu Sharma, Springer   |